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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å R factor = 0.069 wR factor = 0.143 Data-to-parameter ratio = 13.3

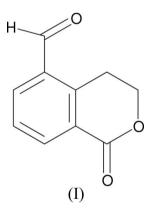
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, erythrocentaurin, $C_{10}H_8O_3$, is a furocoumarin which was isolated from *Enicostema hyssopifolium*. The crystal structure is stabilized by intramolecular $C-H\cdots O$ hydrogen bonds.

5-Formyl-2,3-dihydroisocoumarin

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Comment

The title compound, erythrocentaurin, (I), was isolated from Enicostema hyssopifolium (Willd.) Verdoon of Gentianaceae, which is widely distributed in Southern Pakistan (Omer et al., 1995 and 1996). This plant is considered as medicinally important and used locally by the indigenous people of Tharparkar as a remedy for malaria. In different regions of Pakistan, other species from the same family are used as medicinal plants; they are used as digestive aids, stomachic tonics and also for their depurative, sedative and antipyretic effects (Newall et al., 1996; Sastri, 1952). Chemical surveys reveal that E. hyssopifolium contains alkaloids, flavones and their derivatives (Chaudhri et al., 1975; Ghosal et al., 1974; Popov & Marekov, 1959). Erythrocentaurin, (I), has also been found to be an active agent against serine proteases such as chymotrypsin and trypsin; these proteases are involved in the destruction of certain fibrous proteins (Starkey, 1977). We report here the X-ray crystal structure of erythrocentaurin (I).



The bond lengths in compound (I) show normal values (Allen *et al.*, 1987). The pyrone ring is in a twist-boat conformation, with puckering parameters Q = 0.548 (2) Å, $\theta = 24.8$ (2)° and $\varphi = 272.5$ (5)° (Cremer & Pople, 1975). The formyl group is nearly coplanar with the attached benzene ring $[O3-C10-C1-C8 = 6.3 (5)^{\circ}]$.

The intramolecular C7–H7B···O3 hydrogen bond generates a ring of graph-set motif S(6) (Bernstein *et al.*, 1995). The stabilization of the structure is supported by this intramolecular C–H···O hydrogen bond. A view of the molecular packing along the *c* axis is shown in Fig. 2.

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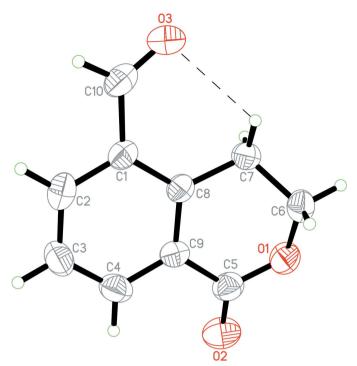


Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. The hydrogen bond is shown as a dashed line.

Experimental

Air-dried plants of E. hyssopifolium (2.5 kg) were chopped and soaked in methanol for a period of 30 d at room temperature. The combined methanol extract was concentrated to yield a crude methanol extract (300 g). This was suspended in water (1 l) and the suspension was further extracted with n-hexane (175 g, 3 l), CHCl₃ (50 g, 3 l) and *n*-butanol (40 g, 3 l). The CHCl₃-soluble fraction was chromatographed on a silica gel column using hexane-CHCl₃; the polarity was increased gradually to afford thirteen fractions. Fraction 5 was submitted to repeated FC (230-400 mesh) and eluted with CHCl₃:*n*-hexane (30:70) to afford the title compound, (I). An $R_{\rm f}$ value of 0.50 was noted on thin layer chromatography (same solvent mixture) and the compound was recrystallized from chloroform (m.p. 413-414 K).

Crystal data

$C_{10}H_8O_3$	$D_x = 1.448 \text{ Mg m}^{-3}$
$M_r = 176.16$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 2249
a = 7.651 (3) Å	reflections
b = 4.0197 (14) Å	$\theta = 1.5 - 25.0^{\circ}$
c = 26.701 (9) Å	$\mu = 0.11 \text{ mm}^{-1}$
$\beta = 100.337 (9)^{\circ}$	T = 293 (2) K
V = 807.9 (5) Å ³	Needle, colorless
Z = 4	$0.37 \times 0.11 \times 0.07 \text{ mm}$
Data collection	
Siemens SMART CCD area-	1573 independent reflections
detector diffractometer	1151 reflections with $I > 2\sigma(I)$
ω scans	$R_{\rm int} = 0.031$
Absorption correction: multi-scan	$\theta_{\rm max} = 26.0^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -6 \rightarrow 9$
$T_{\min} = 0.961, T_{\max} = 0.993$	$k = -4 \rightarrow 4$
44.55	1 00 00

 $-32 \rightarrow 32$

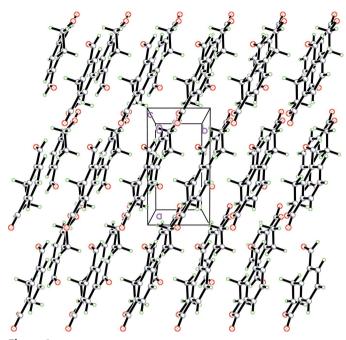


Figure 2 Molecular packing of (I), viewed along the c axis.

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0527P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.069$	+ 0.2194P]
$wR(F^2) = 0.143$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.17	$(\Delta/\sigma)_{\rm max} < 0.001$
1573 reflections	$\Delta \rho_{\rm max} = 0.19 \ {\rm e} \ {\rm \AA}^{-3}$
118 parameters	$\Delta \rho_{\rm min} = -0.23 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	

Table 1			
Hydrogen-bond	geometry (Å,	°).

, ,					
$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$	
$C7-H7B\cdots O3$	0.97	2.35	2.902 (3)	115	

All H atoms were positioned geometrically and allowed to ride on their parent C atoms, with $Csp^2 - H = 0.93$ Å and methylene C-H = 0.96 Å; $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2003).

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